Akashi College			Year	Year 2023		Cours	se I	ntroduction to Nano Jaterials Design			
Course Information											
Course Co	ode	5004			Course Categor	ry General / F		Elective			
Class For	Class Format Lecture				Credits	Aca	Academic Credit: 2				
Departme	Department Mechanie Engineer		cal and Electronic System		Student Grade	Adv	Adv. 1st				
Term First Sem			nester		Classes per We	ek 2					
Textbook Teaching	and/or Materials	Handouts	5								
Instructor		NAKANIS	HI Hiroshi								
Course Objectives											
Objectives are to: Evaluation 1: Understand the various laws that govern the natural world and learn the methods in applying the laws to nanomaterials design through the lectures. Evaluation 2: Deepen one's understanding of quantum mechanics and develop presentation skills in expressing one's opinions and ideas to others plainly through exercises and a presentation. Evaluation 3: Develop the basic skills in applying and expanding nanomaterials design to researches in one's major field. (D, E, H)											
RUDIIC					1						
			Ideal Level of A	Achievement Standard Level of Ach		of Achieve	ment	Achievement)			
Evaluation 1			The student cle and explains th design method	The student clearly understands and explains the nanomaterials design methods.		scribes that ties come from echanics.		The student did not describe that material properties come from the quantum mechanics and did not explain the nanomaterials design methods.			
Evaluation 2			The student clearly understands and explains how to utilize the guantum mechanic algebra.		tilizes the anics algeb	ıra.	The student did not utilize the quantum mechanics algebra.				
Evaluation 3			The student and nanomaterials developing her,	pplies the design for /his field.	The student pr application of t design in her/h	student proposes the lication of the nanomate gn in her/his field.		The student did not propose the application of the nanomaterials design in her/his field.			
Assigned Department Objectives											
Teachin	a Metho	d	-								
Outline Nanomaterials design is a method of designing various materials that support the present and futu and technologies. An objective of this course is to develop a scientific way of thinking by learning nanomaterial design. First, students are going to learn the outline of quantum mechanics, which ex motions of nuclei and electrons that make up a material. Second, the students are going to learn h quantum mechanics clarifies the composition and characteristics (physical properties) of materials. students are going to learn the state-of-the-art nanomaterials design method to design highly-func materials, which will be required in various engineering fields in the future.							the present and future science thinking by learning n mechanics, which explains the s are going to learn how operties) of materials. Lastly, the to design highly-functional				
Style Outline and necessary subjects will be illustrated through theory lectures, followed by practice lecture The student is expected to solve the practice problems with her/his own hands, and to explain her/his solutions to other students easy to understand.							owed by practice lectures. ds, and to explain her/his				
Notice In this course, the learning time guaranteed in the class and the total of the standard self-study time necessary for the preparation / review are 90 hours of study content.								standard self-study time			
Charact	eristics of	of Class /	Division in Le	arning	¥						
 ☑ Active Learning 			□ Aided by IC	T.	☑ Applicable to	o Remote C	lass	Instructor Professionally Experienced			
					1						
Course	Plan										
		-	Theme			Goals					
1st Semeste r	1st Quarter	1st	Outline of Quantu Learn the outlind differences betwe Newtonian mecha mechanics.	Im Mechanics (Firs e of quantum mech en quantum mech anics by comparing	st Half) chanics and nanics and g the two	The student explains the differences between quantum mechanics and Newtonian mechanics					
		2nd I	Dutline of Quantu Learn the methoo mechanically.	um Mechanics (Second Half) d of expressing motions quantum p		The student explains the description of the particle motion in quantum mechanics.					
		3rd	Basics of Quantur Algebra) Learn operator al earn quantum m	n Mechanics 1 (Operator gebra, which is necessary to echanics		The student handles the basic algebra necessary in quantum mechanics.					
		4th	Basics of Quantur Equation) Schrodinger wave n quantum mech equation.	n Mechanics 2 (So e equation is the b anics. Learn Schro	The students explains the relation between wave packet and particle motion.						
		5th	Basics of Quantur Relations I: Coorc Learn the commu coordinates and r	m Mechanics 3 (Co dinates and Mome Itation relation bel nomentum.	The students operates the commutator brackets coordinates and momentum.						
		6th	Basics of Quantur Relations II: Angu Learn the commu angular momentu	ics of Quantum Mechanics 4 (Commutation ations II: Angular Momentum) rn the commutation relation regarding an ular momentum			The students operates the commutator brackets to coordinates and momentum.				

			7th	Basics of Quantur Operators) Learn about Herm	Mechanics 5 (He	ermitian	The student explains the Hermitian, and calculates the time evolution of expectation value of physical quantity.			
			8th	Basics of Quantum Mechanics 6 (Square well Potential) Learn the quantum states of a particle bound by a square-well potential.			The student derives the quantum states of a particle bound by a square-well potential.			
			9th	Basics of Quantum Mechanics 7 (One-Dimensional Scattering Problem and Tunnel Effect) Learn about scattering problems and understand the tunnel effects.				ves the transmiss are-well potential	ion probability energy barrier.	
			10th	Basics of Quantum Mechanics 8 (Harmonic Oscillators) Learn about the quantum states of harmonic oscillators.			The student derives the quantum states of Harmonic Oscillator.			
			11th	Basics of Quantum Mechanics 9 (Lattice Specific Heat) Learn about Einstein solid.				ives the heat capacity of Einstein		
			12th	Electron Configuration of Atom 1 Learn about the quantum states of an electron bounded by the Coulomb force.				ains the quantum m.	states of an	
	2nd Quarte	er	13th	Electron Configuration of Atom 2 (Spin and Quantum Statistics) Learn about the existence of spin, the outline of the quantum statistics, and the periodic laws of elements.			The student explains the electron configuration in an atom.			
			14th	Cohesion Mechanism of atoms in materials (Ionic Bond, Covalent Bond and Metallic Bond) Learn the cohesion mechanisms of atoms in materials. Density Functional Theory and Computational Material Design Learn the density functional theory, the first principle calculation based on the density functional theory, and nanomaterials design using the first-principle calculations.			The student explains the ionic bond, covalent bond and metallic bonds) Learn the cohesion mechanisms of atoms in materials.			
			15th				The student explains the nanomaterials design methods.			
			16th	Term-end examination						
Evaluation Method and Weight (%)										
E		Examination		Practice & Presentation	Mutual Evaluations between students	Behavior	Portfolio	Other	Total	
Subtotal		80		20	0	0	0	0	100	
Basic Ability		20		5	0	0	0	0	25	
Technical Ability		50		5	0	0	0	0	55	
Interdisciplinar y Ability		10		10	0	0	0	0	20	