Tsuyama College		Year	Year 2021		Course Title	Quantum Science						
Course Information												
Course Code 0171				Course Category	Specialize	Specialized / Elective						
Class Format	Lecture			Credits	Academic	Academic Credit: 2						
Department	t Department of Integrated Science and t Technology Communication and Informations System Program			Student Grade	5th	5th						
Term	Year-round			Classes per Week	1	1						
Textbook and/or Teaching Materials	Textbook :	Yasuo Hara an	d Makoto Okazaki	,"Modern Physics for Engineering" (Shokabo)								
Instructor SASAI Yuji												
Course Objectives												
Learning purposes : Quantum mechanics governs the properties of the atoms and molecules that make up the materials of devices that are used daily in modern life. In this subject, students will learn the duality of wave nature and particle nature and the old quantum theory. Then, understand the basics of quantum mechanics as wave mechanics and the quantum number of atoms.												
Course Objective : 1. Understand the old quantum theory and solve related problems. 2. Understand the basics of quantum mechanics as wave mechanics and quantum numbers, and solve related problems.												
Rubric												
]	Ideal Level		Standard Level		Unacceptable Level						
Achievement 1	(t	Can create answers to most of the problems dealt with in class about the old quantum theory.		Can create answers to the problems dealt with in class about the old quantum theory.		Has not reached the left.						
Achievement 2		Can create answers to most of the problems dealt with in class about the guantum mechanics.		Can create answers to the problems dealt with in class about the quantum mechanics.		Has not reached the left.						
Assigned Departr	nent Obje	ctives										
Teaching Method												
	General or Specialized : Specialized											
	Field of learning + Physics											
	Required, Elective, etc. : Must complete subjects											
	Basic disciplines: Mathematical science / physics / general physics											
Outline	Relationship with Educational Objectives : This subject corresponds to the learning objective of each engineering department, "(1) Acquire knowledge about natural science subjects centered on mathematics and physics, and acquire the ability to apply it as basic knowledge about each engineering."											
	Relationship with JABEE programs : The main goal of learning or education in this subject is "(A) Deepening of basic knowledge about technology, A-1: Acquiring knowledge in a wide range of natural sciences as basic knowledge about engineering, and can be explained. "											
	Class outline : Quantum mechanics is important as a basic principle of chemistry and electronic engineering. In this course, student will understand the basics of quantum mechanics as wave mechanics.											
	Course method: Lecture-style lessons will be conducted and exercises will be conducted as appropriate. In the exercise, students will be asked to write a board and explain the answers. Impose an assignment report and proceed with the lesson while confirming the degree of understanding of the students.											
Style	Grado ovalu	ation mothod	5 5	5								
	Exams (60%) + Exercises (40%). Supplementary classes and re-taking exams will be imposed on those with poor grades, and the results of the regular exam will be replaced with a maximum of 60 points.											
	Precautions on the enrollment : This subject is a "subject that requires study outside of class hours". Classes are offered for 15 credit hours per credit, but 30 credit hours are required in addition to this. Follow the instructions of teacher for these studies.											
Notice	Course advice : Read the textbook well. Also, be sure to submit the assignment report by the deadline.											
	Basic subjects : General Physics (3rd year), Differential and Integral I (2), Differential and Integral II (3), Fundamental Differential Equations (3)											
	Related subjects : Electromagnetism (4), Modern Physics (4), Analytical Mechanics(4), Condensed Matter Physics (4), Mathematics subject											
	Attendance advice : Calculate and understand the mathematical formulas. If students are operating e-mail etc. during class, may be asked to leave the room. If student join the class starts within 25 minutes, it will be lateness, and 3 times lateness will result in 1 absence.											
Characteristics of	Class / D	ivision in Le	arning									
Active Learning		Aided by IC	Т	□ Applicable to F	Remote Class	Instructor Professionally Experienced						

Course Plan												
			٦	Theme			Goals					
1st Semeste r		1st	: 0	 Other than math and physics programs: Guidance Mathematics and Physics Program: Guidance 			Guidance					
		2no	d [Duality of light			Understand the photoelectric effect and Compton scattering.					
	1st Quarter	3rc	1 C	Duality of electron			Understand wavefunction, probability density, de Broglie wavelength, and uncertainty principle.					
		er 4th	n s	Schrödinger equation			Understand the derivation of the Schrödinger equation.					
		5th	n 5	Stable state I			Infinitely deep well-shaped potential					
		6th	า ร	Stable state II			General potential, harmonic oscillator potential					
		7th	ר ו	Tunnel effect			Tunnel effect and laser					
		8th	נ ו	1st term midterm exam (above content)			Requires a score of 60 points or higher.					
		9th	n F	Return of answers for the 1st term midterm exam. exam commentary.			Review.					
		10	th A	Angular momentum in quantum mechanics			Qualitative understanding of angular momentum and three-dimensional Schrodinger equation					
		11	th ŀ	Hydrogen atom			Hydrogen atom and spin					
	2nd	12	th (Quantum mechanics of multi-particle system			Schrödinger equation of two-particle system, wave function of identical particles					
	Quarte	er 131	th F	Periodic law of atoms			Atomic shell model and periodic law of atoms					
		14	th F	Fermi distribution and Bose distribution			Fermions and bosons, Pauli principle, chemical potential					
		15	th 1	1st term final exam (contents after the first term mid-term exam)			Requires a score of 60 points or higher.					
		16	th F	Return of answers exam commentary	for the 1st term	final exam.	Review.					
Evaluati	Evaluation Method and Weight (%)											
		Examination		Presentation	Mutual Evaluations between students	Behavior	Portfolio	Other	Total			
Subtotal		60		0	0	0	40	0	100			
Basic Proficiency		30		0	0	0	20	0	50			
Specialized Proficiency		30		0	0	0	20	0	50			